











currently available. They were developed by balanced committees representing all concerned interests. They specify procedures, instrumentation, equipment operating requirements, calculation methods, and uncertainty analysis.

When tests are run in accordance with a Code, the test results themselves, without adjustment for uncertainty, yield the best available indication of the actual performance of the tested equipment. ASME Performance Test Codes do not specify means to compare those results to contractual guarantees. Therefore, it is recommended that the parties to a commercial test agree **before starting the test and preferably before signing the contract** on the method to be used for comparing the test results to the contractual guarantees. It is beyond the scope of any Code to determine or interpret how such comparisons shall be made.

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*inch*:  $\frac{1}{12}$  of a foot, originally established by statute, to be that of three grains of barley, dry and round, placed end-to-end lengthwise.

*joule*: the unit of work or energy equal to the work done by a force of one newton when the point at which the force is applied is displaced one meter in the direction of the force. It is also known as a newton-meter of energy. It is also practically equivalent to the energy expended by an electric current of one ampere flowing for one second through a resistance of one ohm.

*meter*: a unit of length in the SI system, originally intended to be one ten-millionth of the distance along a meridian on Earth from the equator to the pole. Also proposed as the length of a pendulum with a period of one second at sea level and 45 deg N latitude (the middle of France).

*power, drawbar*: drawbar power is:

$$P_D = FV/B$$

where

$$\begin{aligned} P_D &= \text{drawbar power, hp [W]} \\ F &= \text{pull at drawbar, lbf [N]} \\ V &= \text{speed, ft/sec [m/s]} \\ B &= 550 \text{ (ft-lbf)/(hp-sec) [W/W]} \\ &= [1.000 \text{ for Watts}] \end{aligned}$$

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*temperature*: the property of an object which determines the direction of heat flow when that object is placed in thermal contact with another object at a different temperature.

*yard*: a unit of length containing 36 in.; it is the distance, at 62°F, between two transverse lines in gold plugs set in a bronze bar, called the British

Imperial Yard and kept at the Standards Office of the Board of Trade at Westminster in Great Britain.

### 3.3 DEFINITIONS FOR USE

The following definitions are approved for use in ASME PTCs.

*air*: the natural atmospheric mixture of nitrogen, oxygen, water vapor, carbon dioxide, argon, and small quantities of other rare gases. The psychrometric properties of air are given in ASHRAE's *Brochure on Psychrometry*, in which the reference conditions are 59°F and 29.921 in.Hg.

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Methods of measurement of atmospheric pressure are contained in ASME PTC 19.2, *Pressure Measurement*. All corrections required by ASME PTC 19.2 in reducing observations to standard conditions must be followed in all ASME PTC calculations.

The International Standard Atmosphere is 760 mm of mercury at the standard gravity of 9.806 65 m/s<sup>2</sup> at 0°C which is equal to 29.921 in. of mercury at the standard gravity of 32.174 06 ft/sec<sup>2</sup> at 32°F. In U.S. Customary units, this is equal to 14.695 95 lbf/in.<sup>2</sup>. In the SI system, this is equal to 101.325 kPa.

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*yard*: a unit of length containing 36 in.; it is the distance, at 62°F, between two transverse lines in gold plugs set in a bronze bar, called the British

Imperial Yard and kept at the Standards Office of the Board of Trade at Westminster in Great Britain.

### 3.3 DEFINITIONS FOR USE

The following definitions are approved for use in ASME PTCs.

*air*: the natural atmospheric mixture of nitrogen, oxygen, water vapor, carbon dioxide, argon, and small quantities of other rare gases. The psychrometric properties of air are given in ASHRAE's *Brochure on Psychrometry*, in which the reference conditions are 59°F and 29.921 in.Hg.

*atmospheric pressure, standard*: when pressure is measured in atmospheres it is measured in terms of units which approximate the vertical air pressure on a horizontal unit area at sea level. This unit then becomes the weight of the superincumbent atmosphere, assuming that the atmosphere is at rest.

Methods of measurement of atmospheric pressure are contained in ASME PTC 19.2, *Pressure Measurement*. All corrections required by ASME PTC 19.2 in reducing observations to standard conditions must be followed in all ASME PTC calculations.

The International Standard Atmosphere is 760 mm of mercury at the standard gravity of 9.806 65 m/s<sup>2</sup> at 0°C which is equal to 29.921 in. of mercury at the standard gravity of 32.174 06 ft/sec<sup>2</sup> at 32°F. In U.S. Customary units, this is equal to 14.695 95 lbf/in.<sup>2</sup>. In the SI system, this is equal to 101.325 kPa.

For other relationships of the International Standard Atmosphere, ASME PTC 19.2 should be consulted. When the barometric pressure is not stated, it is understood that a vacuum below atmosphere or a gage pressure above atmosphere is based on the standard atmosphere of 760 mm of mercury at 0°C.

*bulk modulus of elasticity (E)*: a measure of the fluid's compressibility. That compressibility is measured in terms of the fluid's decrease in volume,  $\Delta V$ , compared to its original volume,  $V$ , when subjected to an incremental pressure change,  $\Delta p$ .  $E$  is path- or process-dependent. Thus,

$$E_N = \ell_{\text{im}} \left\{ - \left[ \frac{\Delta p}{\left( \frac{\Delta V}{V} \right)} \right]_N \right\} = -V \left( \frac{\partial p}{\partial V} \right)$$

where

$N$  = the path or process

*mercury, density of:* density of mercury at 32°F [0°C] is 0.491 154 lbm/in<sup>3</sup> [13 595.1 kg/m<sup>3</sup>]. Density at other temperatures may be computed from the equation:

$$\rho_t = 13\,595.1 / (1 + 1.184\,56 \times 10^{-4} t + 9.205 \times 10^{-9} t^2 + 6.608 \times 10^{-12} t^3 + 6.732 \times 10^{-14} t^4)$$

where

$t$  = temperature of mercury, °C

$\rho_t$  = density of mercury at temperature  $t$ , kg/m<sup>3</sup>

*sound:* acoustic output of a machine or process is its sound power ( $W$ ). This is usually expressed as a sound power level in decibels. Sound power cannot be measured directly. Instead, sound pressures are measured and sound powers calculated therefrom. Modern instrumentation has imbedded this computational capability. Sound pressures ( $p$ ) are usually expressed as sound pressure levels, also in decibels. A decibel is the logarithmic expression of the ratio of a quantity to a particular reference quantity. The reference quantity for sound power levels ( $W_o$ ) is 1  $\mu$ W and that for sound pressure levels ( $p_o$ ) is 20  $\mu$ Pa, hence

sound power level (dB) = 10  $\log_{10}$  ( $W/W_o$ )

sound pressure level (dB) = 20  $\log_{10}$  ( $p/p_o$ )

*output, net:* of an engine and generator unit, defined by the formula:

$$\text{Net output (kW)} = \left( \begin{array}{c} \text{electrical power} \\ \text{output of} \\ \text{generator (kW)} \end{array} \right) - \left( \begin{array}{c} \text{auxiliary} \\ \text{power} \\ \text{supplied (kW)} \end{array} \right)$$

Auxiliary power supplied is that external power necessary for the unit's operation and includes, but is not limited to, excitation power, power for separately driven lube oil pumps, hydraulic oil pumps, generator cooling water pumps, fans for generator ventilation, and seal evacuators.

*power, air or gas:* the equations for air power invoke a compressibility factor  $K_p$  in this fashion

$$P = \left\{ (p_2 - p_1) Q_1 K_p + m [(\alpha_2 V_2^2 - \alpha_1 V_1^2) / 2g_c + g(Z_2 - Z_1) / g_c] \right\} / A$$

where

$P$  = power, hp [W]

$$K_p = \ell n \frac{p_2/p_1}{\frac{p_2}{p_1} - 1}$$

(for a reversible isothermal process)

$$K_p = \frac{\gamma}{\gamma - 1} \left[ \left( \frac{p_2}{p_1} \right)^{\frac{\gamma-1}{\gamma}} - 1 \right] / \left[ \frac{p_2}{p_1} - 1 \right]$$

(for an isentropic process)

$$K_p = \frac{n}{n - 1} \left[ \left( \frac{p_2}{p_1} \right)^{\frac{n-1}{n}} - 1 \right] / \left[ \frac{p_2}{p_1} - 1 \right]$$

(for a reversible polytropic process)

$$K_p = 1 \text{ [for a reversible isochoric process (constant density)]}$$

$\gamma$  = ratio of specific heats

$n$  = polytropic exponent

$p_1$  = inlet pressure, lb/ft<sup>2</sup> [Pa]

$p_2$  = outlet pressure, lb/ft<sup>2</sup> [Pa]

$Q_1$  = volumetric flow rate, ft<sup>3</sup>/sec [m<sup>3</sup>/s]

$\alpha_1$  = kinetic energy factor at inlet

$\alpha_2$  = kinetic energy factor at outlet

$V_1$  = average velocity at inlet, ft/sec [m/s]

$V_2$  = average velocity at outlet, ft/sec [m/s]

$Z_1$  = elevation at inlet, ft [m]

$Z_2$  = elevation at outlet, ft [m]

$g$  = acceleration due to gravity, ft/sec<sup>2</sup> [m/s<sup>2</sup>]

$g_c$  = conversion factor, 32.174(ft-lbm)/(lb-sec<sup>2</sup>) [1(m-kg)/(N-s<sup>2</sup>)]

$A$  = conversion factor, 550 (ft-lb)/(hp-sec) [1(N-m)/(W-s)]

*power, water:* the energy flux contained in flowing water. Water power is to be computed from the equation

$$P_w = g_p Q (H_1 - H_2) / A = Q (p_1 - p_2) / A$$

where

$P_w$  = water power, hp [W]

$\rho$  = density of water, slug/ft<sup>3</sup> [kg/m<sup>3</sup>]

$Q$  = volumetric flow, ft<sup>3</sup>/sec [m<sup>3</sup>/s]

$H_1$  = higher head, ft [m]

$H_2$  = lower head, ft [m]

$p_1$  = higher pressure, lbf/ft<sup>2</sup> [Pa]

$p_2$  = lower pressure, lbf/ft<sup>2</sup> [Pa]

$A$  = 550 (ft-lbf)/(hp-sec) [1 W/W]

$g$  = local acceleration of gravity, ft/sec<sup>2</sup> [m/s<sup>2</sup>]

All heads and pressures shall be measured at, or corrected to, stagnation conditions.

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$A$  = 550 (ft-lbf)/(hp-sec) [1 W/W]

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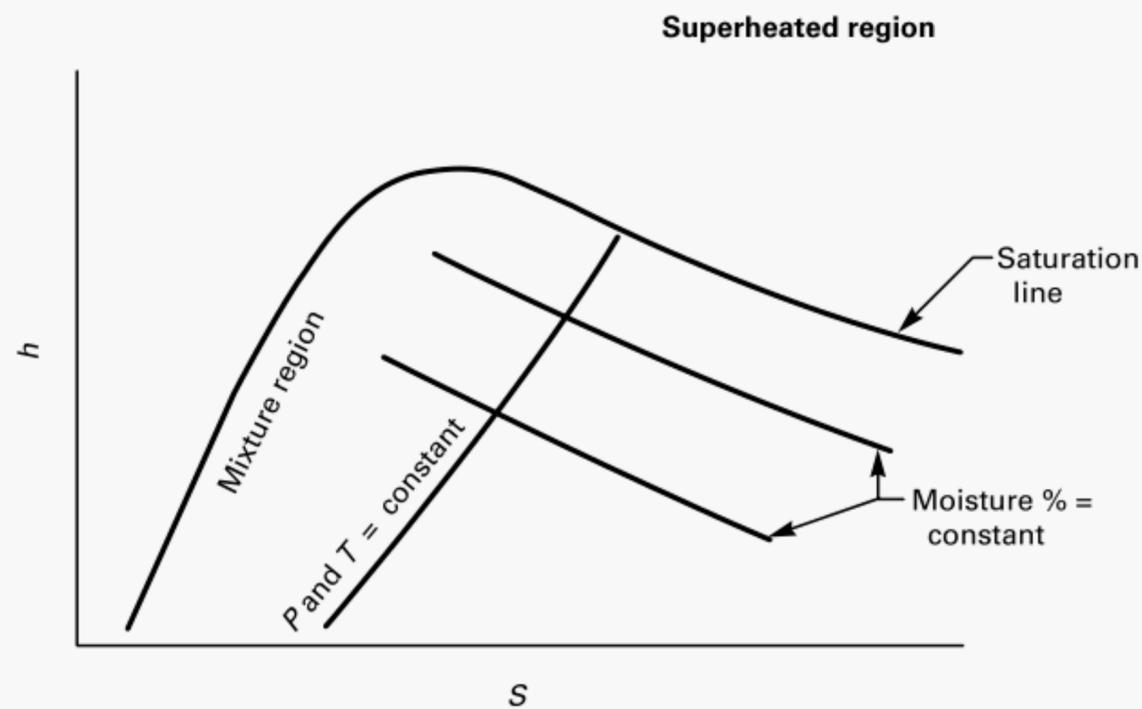


FIG. 3.3 ENTHALPY-ENTROPY DIAGRAM FOR WATER

For ideal pump work, refer to the definition of *pumps* where internal losses are negligible.

*steam point*: is defined as the temperature of an equilibrium mixture of liquid and condensing water vapor at one standard atmospheric pressure (373.15 K).

*steam quality*: inside the liquid-vapor-mixture region, quality  $x$  is defined as the fraction of the mass which is in the vapor state relative to the total mass of the two-phase mixture:

$$x = m_g / (m_g + m_f)$$

For example, if a mixture at saturated pressure and temperature contains 2.5% moisture, its quality is 97.5%. "Wet" steam means its state is in the mixture region wherein its state is defined by either its pressure or temperature and the portion of water substance which is the vapor phase.

Dry and saturated steam exists when all of the mass of water substance is in the vapor phase at saturation pressure and temperature. This condition exists along the "saturation line" which divides the two-phase mixture region from the superheated region ( $x = 100\%$  and moisture = 0%). See Fig. 3.3.

*steam rate*: of an engine, turbine, or complete plant, is expressed as the actual mass of steam per unit of time per unit of output, often expressed in units of lbm/kW.

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*temperature, reference*: the datum for the steam tables is the triple point of water. The 1968 International Practical Temperature Scale establishes this at 0.01°C or 32.018°F.

For gases, the preferred standard temperature for PTC work is 59°F (15°C). Various industries use different reference conditions; caution is advised.

*thermal conductivity*: the coefficient of proportionality in the equation of heat transfer by steady unidirectional conduction proposed by Fourier in 1882:  $q = -k A dT/dx$  where  $q$  is the rate of heat conduction along the  $x$ -axis,  $A$  is the cross-sectional area of the path normal to the  $x$ -axis, and  $-dT/dx$  is the temperature gradient along the path. See Table 5.8. The units of  $k$  are Btu/(hr-ft-°F) [W/(m-°C)]

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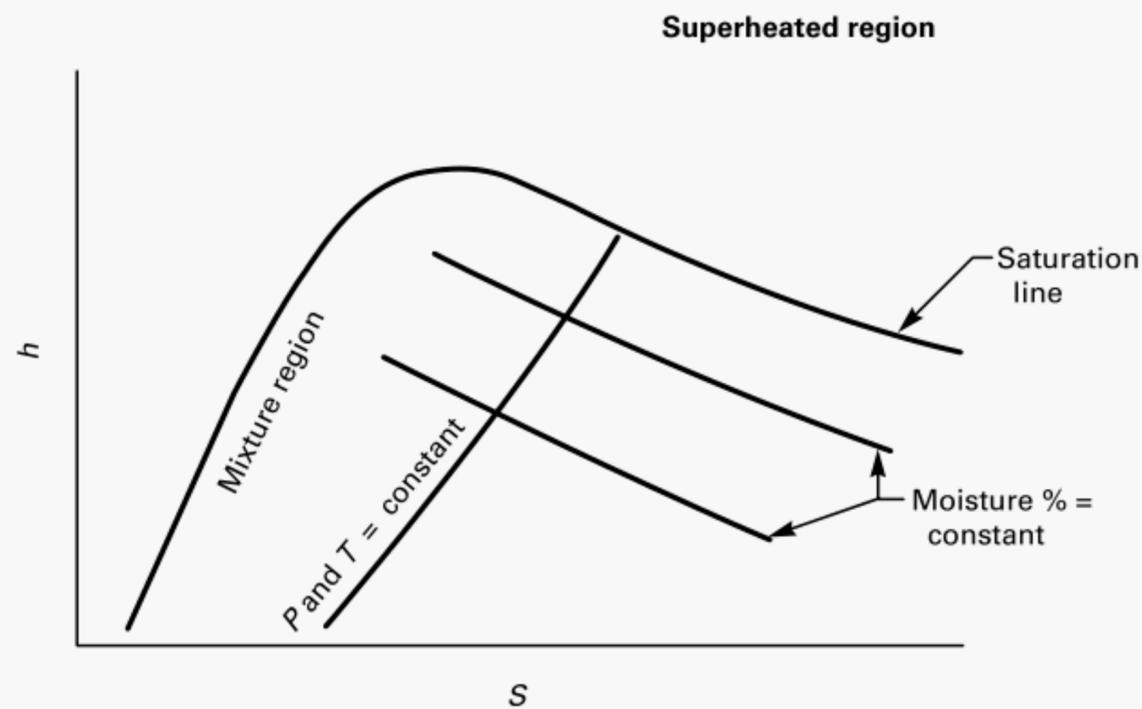


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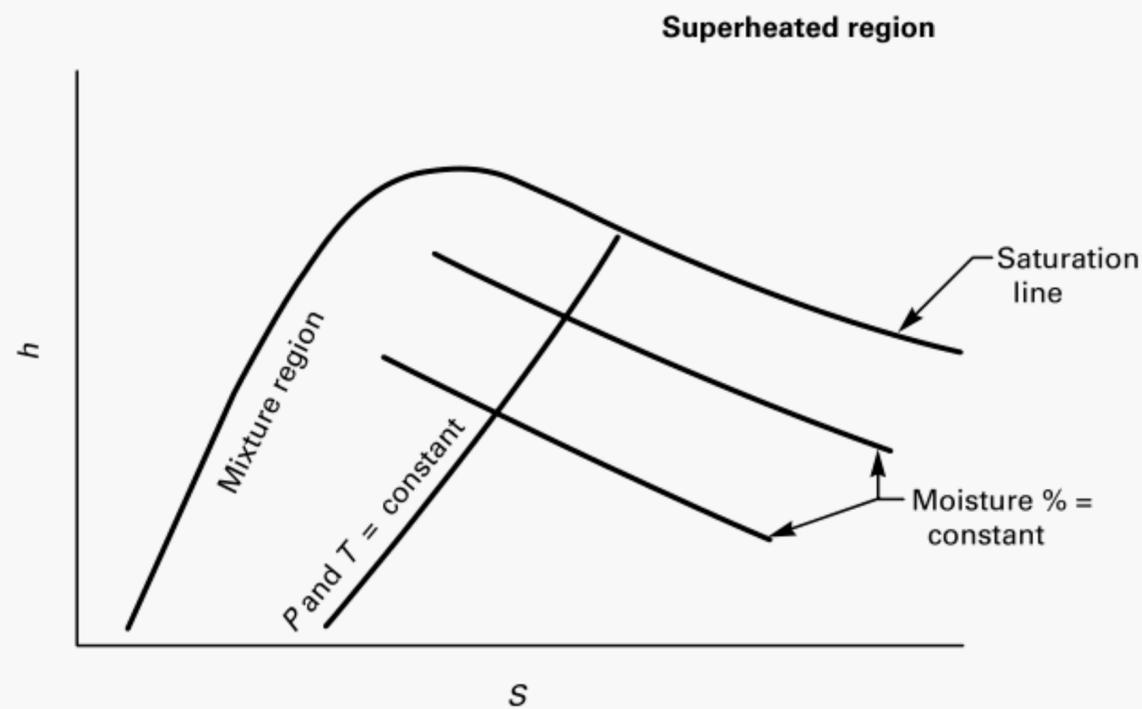


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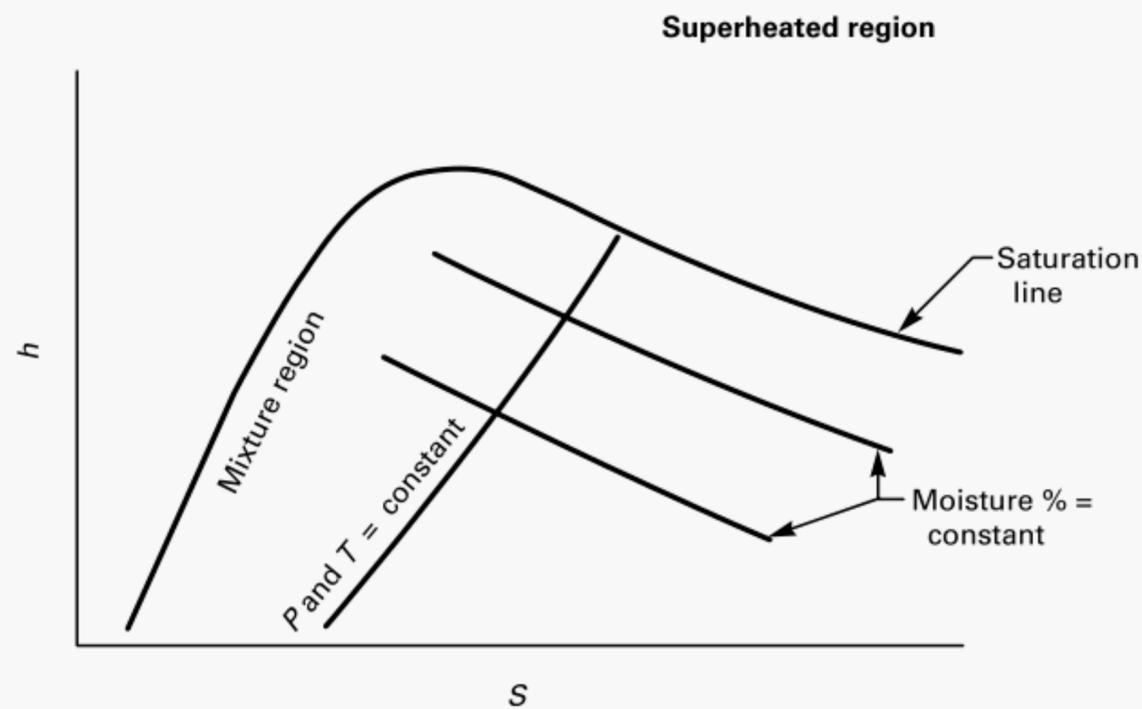


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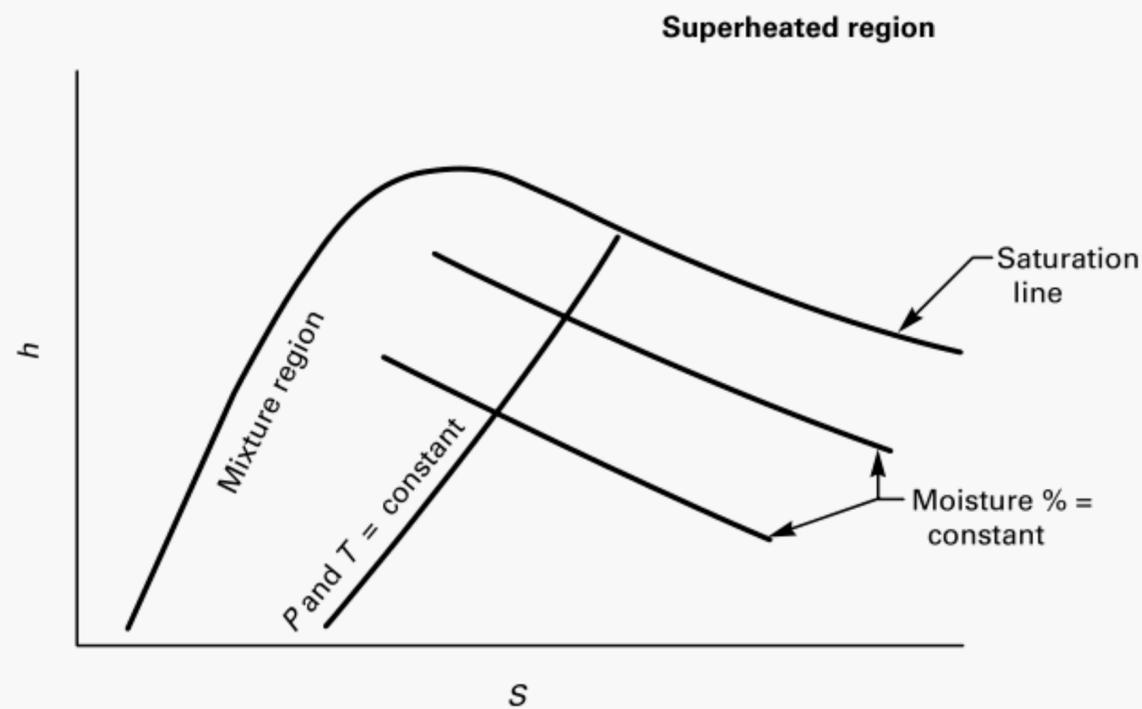


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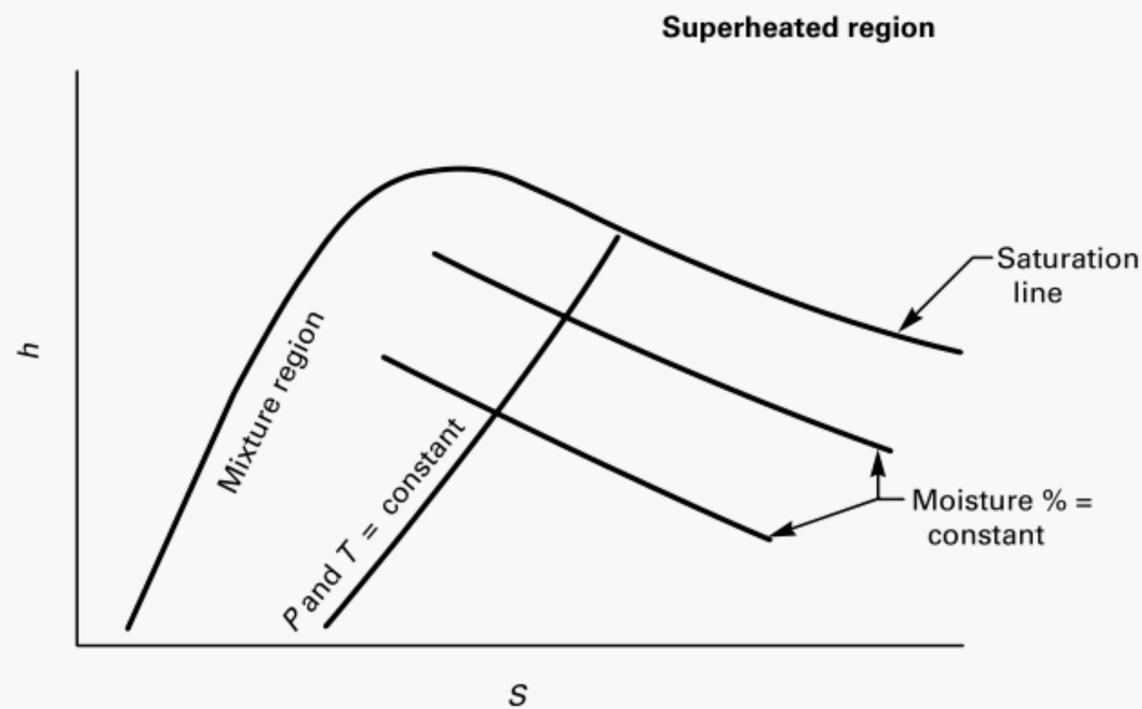


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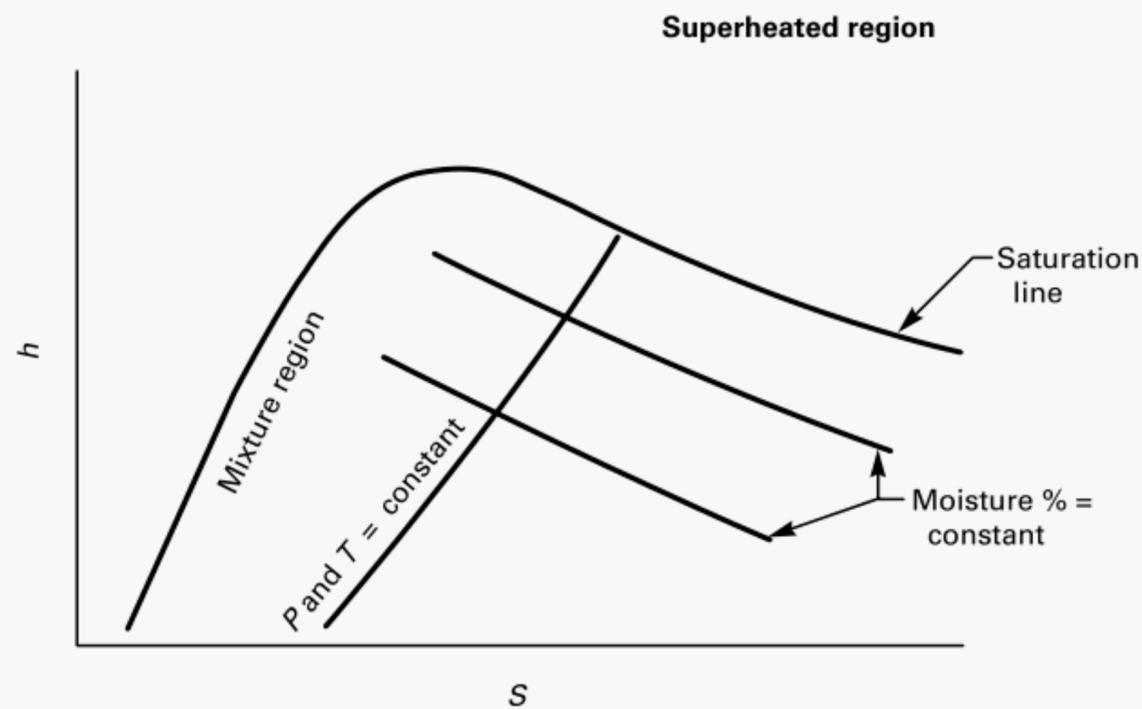


FIG. 3.3 ENTHALPY-ENTROPY DIAGRAM FOR WATER

For ideal pump work, refer to the definition of *pumps* where internal losses are negligible.

*steam point*: is defined as the temperature of an equilibrium mixture of liquid and condensing water vapor at one standard atmospheric pressure (373.15 K).

*steam quality*: inside the liquid-vapor-mixture region, quality  $x$  is defined as the fraction of the mass which is in the vapor state relative to the total mass of the two-phase mixture:

$$x = m_g / (m_g + m_f)$$

For example, if a mixture at saturated pressure and temperature contains 2.5% moisture, its quality is 97.5%. "Wet" steam means its state is in the mixture region wherein its state is defined by either its pressure or temperature and the portion of water substance which is the vapor phase.

Dry and saturated steam exists when all of the mass of water substance is in the vapor phase at saturation pressure and temperature. This condition exists along the "saturation line" which divides the two-phase mixture region from the superheated region ( $x = 100\%$  and moisture = 0%). See Fig. 3.3.

*steam rate*: of an engine, turbine, or complete plant, is expressed as the actual mass of steam per unit of time per unit of output, often expressed in units of lbm/kW.

*temperature, absolute*: the approximate value of the thermodynamic temperature as defined by the International Practical Temperature Scale. In the U.S. customary system, the absolute temperature is expressed in degrees Rankine, and in the SI (metric) system, in kelvin. For relations between these scales, see Section 5.

*temperature, reference*: the datum for the steam tables is the triple point of water. The 1968 International Practical Temperature Scale establishes this at 0.01°C or 32.018°F.

For gases, the preferred standard temperature for PTC work is 59°F (15°C). Various industries use different reference conditions; caution is advised.

*thermal conductivity*: the coefficient of proportionality in the equation of heat transfer by steady unidirectional conduction proposed by Fourier in 1882:  $q = -k A dT/dx$  where  $q$  is the rate of heat conduction along the  $x$ -axis,  $A$  is the cross-sectional area of the path normal to the  $x$ -axis, and  $-dT/dx$  is the temperature gradient along the path. See Table 5.8. The units of  $k$  are Btu/(hr-ft-°F) [W/(m-°C)]

*thermal unit conversions*: using the identities as defined in the latest edition of ASME steam tables (Appendix 4B) and tables of conversion factors, the following conversions are derived:

Identities:

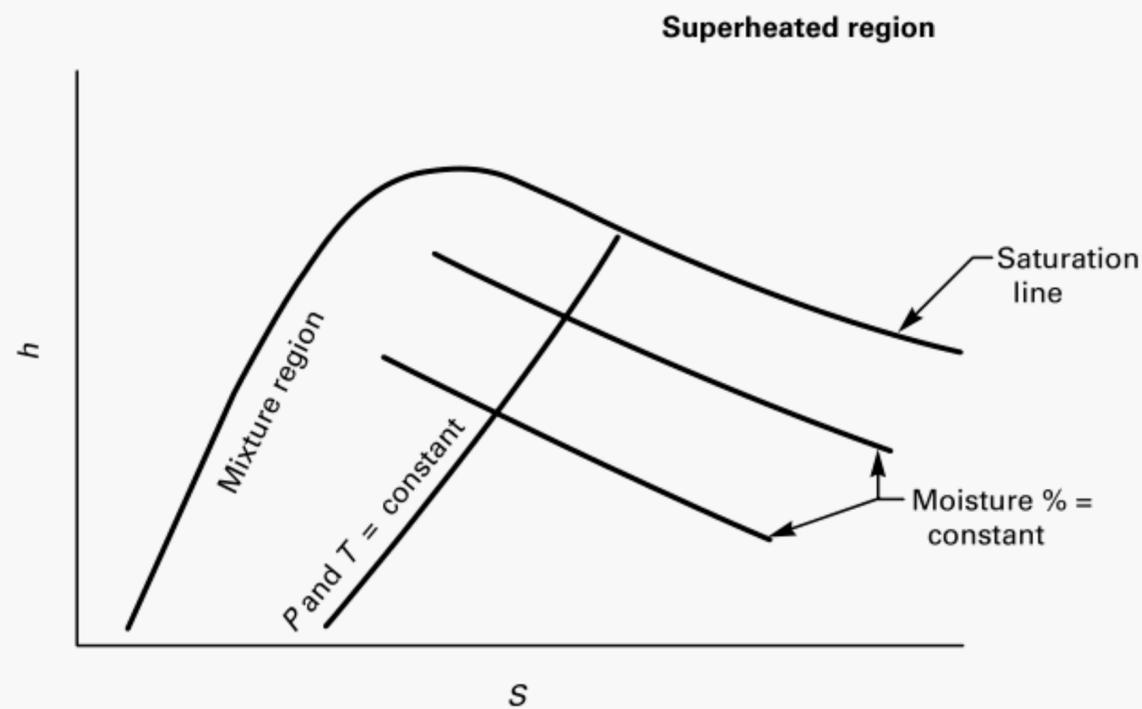


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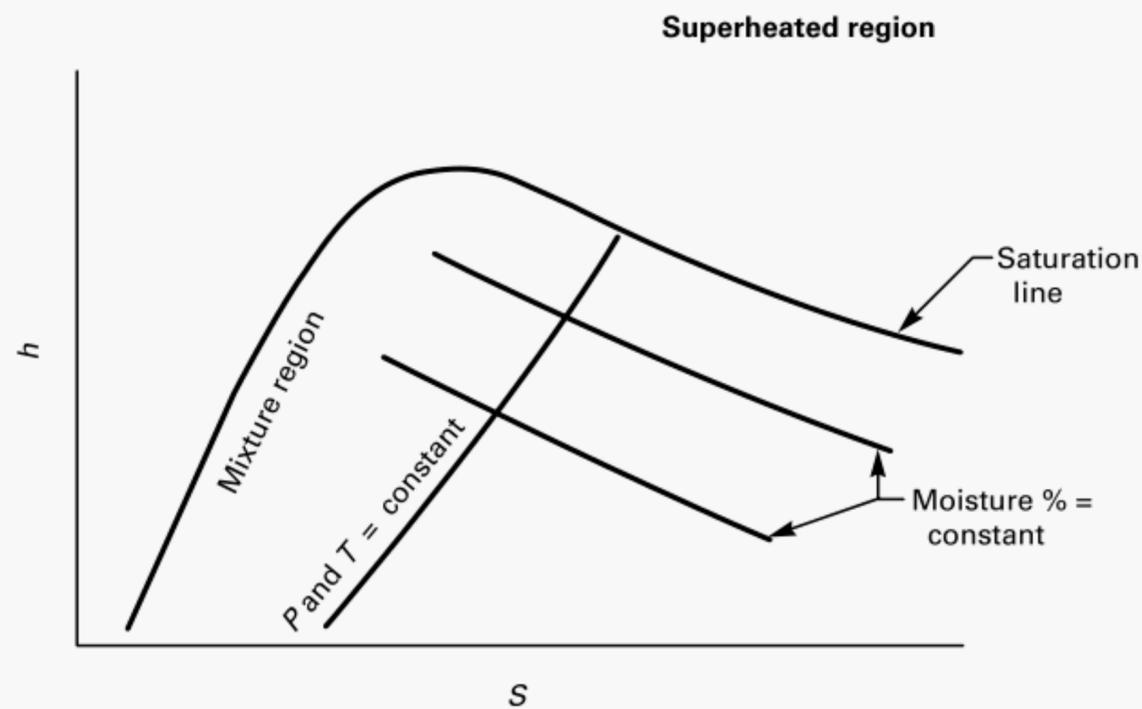


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*thermal unit conversions*: using the identities as defined in the latest edition of ASME steam tables (Appendix 4B) and tables of conversion factors, the following conversions are derived:

Identities:

**TABLE 5.3**  
**CONVERSION FACTORS FOR SPECIFIC VOLUME (volume/mass)**

To obtain Multiply, by ↓ ↘ ↙	$\frac{\text{ft}^3}{\text{lbm}}$	$\frac{\text{in.}^3}{\text{lbm}}$	$\frac{\text{U.S. gal}}{\text{lbm}}$	$\frac{\text{liter}}{\text{kg}}$	$\frac{\text{m}^3}{\text{kg}}$
$\frac{\text{ft}^3}{\text{lbm}}$	1	1 728	$\frac{1\,728}{231}$ = 7.480 519 48	$\frac{30.48^3}{453.592\,37}$ = 62.427 960 6	$\frac{30.48^3 \times 10^{-6}}{0.453\,592\,37}$ = 0.062 427 960 6
$\frac{\text{in.}^3}{\text{lbm}}$	$\frac{1.0}{1\,728}$ = 0.000 578 703 704	1	$\frac{1.0}{231}$ = 0.004 329 004 33	$\frac{2.54^3}{453.592\,37}$ = 0.036 127 292 0	$\frac{2.54^3 \times 10^{-6}}{0.453\,592\,37}$ = 0.000 036 127 292
$\frac{\text{U.S. gal}}{\text{lbm}}$	$\frac{231}{1\,728}$ = 0.133 680 556	231	1	$\frac{231 \times 2.54^3}{453.592\,37}$ = 8.345 404 45	$\frac{231 \times 2.54^3 \times 10^{-6}}{0.453\,592\,37}$ = 0.008 345 404 45
$\frac{\text{liter}}{\text{kg}}$	$\frac{453.592\,37}{30.48^3}$ = 0.016 018 463 4	$\frac{453.592\,37}{2.54^3}$ = 27.679 904 7	$\frac{453.592\,37}{231 \times 2.54^3}$ = 0.119 826 427	1	0.001
$\frac{\text{m}^3}{\text{kg}}$	$\frac{0.453\,592\,37}{30.48^3 \times 10^{-6}}$ = 16.018 463 4	$\frac{0.453\,592\,37}{2.54^3 \times 10^{-6}}$ = 27 679.904 7	$\frac{0.453\,592\,37}{231 \times 2.54^3 \times 10^{-6}}$ = 119.826 427	1000	1

## GENERAL NOTE:

All values given in the rational fractions are exact except 1 U.S. gal = 231 in.<sup>3</sup> (NBS Misc. Pub. 233 P5).

Example: 1 U.S. gal/lbm = 0.133 680 556 ft<sup>3</sup>/lbm

**TABLE 5.3**  
**CONVERSION FACTORS FOR SPECIFIC VOLUME (volume/mass)**

To obtain Multiply, by ↓ ↘ ↙	$\frac{\text{ft}^3}{\text{lbm}}$	$\frac{\text{in.}^3}{\text{lbm}}$	$\frac{\text{U.S. gal}}{\text{lbm}}$	$\frac{\text{liter}}{\text{kg}}$	$\frac{\text{m}^3}{\text{kg}}$
$\frac{\text{ft}^3}{\text{lbm}}$	1	1 728	$\frac{1\ 728}{231}$ = 7.480 519 48	$\frac{30.48^3}{453.592\ 37}$ = 62.427 960 6	$\frac{30.48^3 \times 10^{-6}}{0.453\ 592\ 37}$ = 0.062 427 960 6
$\frac{\text{in.}^3}{\text{lbm}}$	$\frac{1.0}{1\ 728}$ = 0.000 578 703 704	1	$\frac{1.0}{231}$ = 0.004 329 004 33	$\frac{2.54^3}{453.592\ 37}$ = 0.036 127 292 0	$\frac{2.54^3 \times 10^{-6}}{0.453\ 592\ 37}$ = 0.000 036 127 292
$\frac{\text{U.S. gal}}{\text{lbm}}$	$\frac{231}{1\ 728}$ = 0.133 680 556	231	1	$\frac{231 \times 2.54^3}{453.592\ 37}$ = 8.345 404 45	$\frac{231 \times 2.54^3 \times 10^{-6}}{0.453\ 592\ 37}$ = 0.008 345 404 45
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Example: 1 U.S. gal/lbm = 0.133 680 556 ft<sup>3</sup>/lbm

**TABLE 5.5**  
**CONVERSION FACTORS FOR SPECIFIC ENTROPY, SPECIFIC HEAT, AND GAS CONSTANT**  
(energy/mass × temperature)

To obtain → Multiply, by ↗ ↓	$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{\text{kcal}}{\text{g} \times \text{K}}$	$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{\text{kJ}}{\text{kg} \times \text{K}}$
$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	1	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778.169 262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.000 293 071 070	41.868	0.0001	$\frac{4.186\ 8}{9.806\ 65}$ = 0.426 934 784	4.186 8
$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 0.001 285 067 46	1	$\frac{453.592\ 37 \times 30.48}{3.6 \times 10^{13}/980.665}$ = 3.766 160 97 × 10 <sup>-7</sup>	$\frac{30.48 \times 980.665 \times 9/5}{10^6}$ = 0.053 803 204 6	$\frac{980.665 \times 30.48}{2\ 326 \times 10^7}$ = 1.285 067 46 × 10 <sup>-6</sup>	$\frac{30.48 \times 10^{-5} \times 9/5}{0.000\ 548\ 64}$ = 0.000 548 64	$\frac{980.665 \times 30.48 \times 10^{-7}}{5/9}$ = 0.005 380 320 46
$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{3\ 600\ 000}{2.326 \times 453.592\ 37}$ = 3 412.141 63	$\frac{3.6 \times 10^{13}/980.665}{453.592\ 37 \times 30.48}$ = 2 655 223.73	1	$\frac{3.6 \times 10^7 \times 9/5}{453.592\ 37}$ = 142 859.546	$\frac{3\ 600\ 000}{2\ 326 \times 453.592\ 37}$ = 3.412 141 63	$\frac{3.6 \times 10^8 \times 9/5}{980.665 \times 453.592\ 37}$ = 1 456 761 95	$\frac{3.6 \times 10^6 \times 9/5}{453.592\ 37}$ = 14 285.954 6
$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592\ 37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41\ 868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066\ 5}$ = 0.010 197 1621	0.1
$\frac{\text{kcal}}{\text{g} \times \text{K}}$	1000	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778 169.262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.293 071 070	41 868	1	$\frac{4\ 186.8}{9.806\ 65}$ = 426.934 784	4 186.8
$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{9.806\ 65}{4.186\ 8}$ = 2.342 278 11	$\frac{10^5}{30.48 \times 9/5}$ = 1 822.688 83	$\frac{980.665 \times 453.592\ 37}{3.6 \times 10^8 \times 9/5}$ = 0.000 686 453 953	98.066 5	$\frac{9.806\ 65}{4\ 186.8}$ = 0.002 342 278 11	1	9.806 65
$\frac{\text{kJ}}{\text{kg} \times \text{K}}$	$\frac{1.0}{4.186\ 8}$ = 0.238 845 897	$\frac{10^7 \times 5/9}{980.665 \times 30.48}$ = 185.862 535	$\frac{453.592\ 37}{3.6 \times 10^6 \times 9/5}$ = 6.999 882 25 × 10 <sup>-5</sup>	10	$\frac{1.0}{4\ 186.8}$ = 0.000 238 845 897	$\frac{1.0}{9.806\ 65}$ = 0.101 971 621	1

GENERAL NOTE:  
All values given in the rational fractions are exact.

Example: 1 Btu / (lbm × R) = 778.169 262 ft × lbf / (lbm × R)

**TABLE 5.5**  
**CONVERSION FACTORS FOR SPECIFIC ENTROPY, SPECIFIC HEAT, AND GAS CONSTANT**  
 (energy/mass × temperature)

To obtain → Multiply, by ↗ ↓	$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{\text{kcal}}{\text{g} \times \text{K}}$	$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{\text{kJ}}{\text{kg} \times \text{K}}$
$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	1	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778.169 262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.000 293 071 070	41.868	0.0001	$\frac{4.186\ 8}{9.806\ 65}$ = 0.426 934 784	4.186 8
$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 0.001 285 067 46	1	$\frac{453.592\ 37 \times 30.48}{3.6 \times 10^{13}/980.665}$ = 3.766 160 97 × 10 <sup>-7</sup>	$\frac{30.48 \times 980.665 \times 9/5}{10^6}$ = 0.053 803 204 6	$\frac{980.665 \times 30.48}{2\ 326 \times 10^7}$ = 1.285 067 46 × 10 <sup>-6</sup>	$30.48 \times 10^{-5} \times 9/5$ = 0.000 548 64	$\frac{980.665 \times 30.48 \times 10^{-7}}{5/9}$ = 0.005 380 320 46
$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{3\ 600\ 000}{2.326 \times 453.592\ 37}$ = 3 412.141 63	$\frac{3.6 \times 10^{13} / 980.665}{453.592\ 37 \times 30.48}$ = 2 655 223.73	1	$\frac{3.6 \times 10^7 \times 9/5}{453.592\ 37}$ = 142 859.546	$\frac{3\ 600\ 000}{2\ 326 \times 453.592\ 37}$ = 3.412 141 63	$\frac{3.6 \times 10^8 \times 9/5}{980.665 \times 453.592\ 37}$ = 1 456 761 95	$\frac{3.6 \times 10^6 \times 9/5}{453.592\ 37}$ = 14 285.954 6
$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592\ 37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41\ 868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066\ 5}$ = 0.010 197 1621	0.1
$\frac{\text{kcal}}{\text{g} \times \text{K}}$	1000	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778 169.262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.293 071 070	41 868	1	$\frac{4\ 186.8}{9.806\ 65}$ = 426.934 784	4 186.8
$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{9.806\ 65}{4.186\ 8}$ = 2.342 278 11	$\frac{10^5}{30.48 \times 9/5}$ = 1 822.688 83	$\frac{980.665 \times 453.592\ 37}{3.6 \times 10^8 \times 9/5}$ = 0.000 686 453 953	98.066 5	$\frac{9.806\ 65}{4\ 186.8}$ = 0.002 342 278 11	1	9.806 65
$\frac{\text{kJ}}{\text{kg} \times \text{K}}$	$\frac{1.0}{4.186\ 8}$ = 0.238 845 897	$\frac{10^7 \times 5/9}{980.665 \times 30.48}$ = 185.862 535	$\frac{453.592\ 37}{3.6 \times 10^6 \times 9/5}$ = 6.999 882 25 × 10 <sup>-5</sup>	10	$\frac{1.0}{4\ 186.8}$ = 0.000 238 845 897	$\frac{1.0}{9.806\ 65}$ = 0.101 971 621	1

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Example: 1 Btu / (lbm × R) = 778.169 262 ft × lbf / (lbm × R)

**TABLE 5.5**  
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 (energy/mass × temperature)

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$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592\ 37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41\ 868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066\ 5}$ = 0.010 197 1621	0.1
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**CONVERSION FACTORS FOR SPECIFIC ENTROPY, SPECIFIC HEAT, AND GAS CONSTANT**  
 (energy/mass × temperature)

To obtain → Multiply, by ↗ ↓	$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{\text{kcal}}{\text{g} \times \text{K}}$	$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{\text{kJ}}{\text{kg} \times \text{K}}$
$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	1	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778.169 262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.000 293 071 070	41.868	0.0001	$\frac{4.186\ 8}{9.806\ 65}$ = 0.426 934 784	4.186 8
$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 0.001 285 067 46	1	$\frac{453.592\ 37 \times 30.48}{3.6 \times 10^{13}/980.665}$ = 3.766 160 97 × 10 <sup>-7</sup>	$\frac{30.48 \times 980.665 \times 9/5}{10^6}$ = 0.053 803 204 6	$\frac{980.665 \times 30.48}{2\ 326 \times 10^7}$ = 1.285 067 46 × 10 <sup>-6</sup>	$\frac{30.48 \times 10^{-5} \times 9/5}{0.000\ 548\ 64}$ = 0.000 548 64	$\frac{980.665 \times 30.48 \times 10^{-7}}{5/9}$ = 0.005 380 320 46
$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{3\ 600\ 000}{2.326 \times 453.592\ 37}$ = 3 412.141 63	$\frac{3.6 \times 10^{13} / 980.665}{453.592\ 37 \times 30.48}$ = 2 655 223.73	1	$\frac{3.6 \times 10^7 \times 9/5}{453.592\ 37}$ = 142 859.546	$\frac{3\ 600\ 000}{2\ 326 \times 453.592\ 37}$ = 3.412 141 63	$\frac{3.6 \times 10^8 \times 9/5}{980.665 \times 453.592\ 37}$ = 1 456 761 95	$\frac{3.6 \times 10^6 \times 9/5}{453.592\ 37}$ = 14 285.954 6
$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592\ 37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41\ 868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066\ 5}$ = 0.010 197 1621	0.1
$\frac{\text{kcal}}{\text{g} \times \text{K}}$	1000	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778 169.262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.293 071 070	41 868	1	$\frac{4\ 186.8}{9.806\ 65}$ = 426.934 784	4 186.8
$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{9.806\ 65}{4.186\ 8}$ = 2.342 278 11	$\frac{10^5}{30.48 \times 9/5}$ = 1 822.688 83	$\frac{980.665 \times 453.592\ 37}{3.6 \times 10^8 \times 9/5}$ = 0.000 686 453 953	98.066 5	$\frac{9.806\ 65}{4\ 186.8}$ = 0.002 342 278 11	1	9.806 65
$\frac{\text{kJ}}{\text{kg} \times \text{K}}$	$\frac{1.0}{4.186\ 8}$ = 0.238 845 897	$\frac{10^7 \times 5/9}{980.665 \times 30.48}$ = 185.862 535	$\frac{453.592\ 37}{3.6 \times 10^6 \times 9/5}$ = 6.999 882 25 × 10 <sup>-5</sup>	10	$\frac{1.0}{4\ 186.8}$ = 0.000 238 845 897	$\frac{1.0}{9.806\ 65}$ = 0.101 971 621	1

GENERAL NOTE:  
 All values given in the rational fractions are exact.

Example: 1 Btu / (lbm × R) = 778.169 262 ft × lbf / (lbm × R)

**TABLE 5.5**  
**CONVERSION FACTORS FOR SPECIFIC ENTROPY, SPECIFIC HEAT, AND GAS CONSTANT**  
 (energy/mass × temperature)

To obtain → Multiply, by ↗ ↓	$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{\text{kcal}}{\text{g} \times \text{K}}$	$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{\text{kJ}}{\text{kg} \times \text{K}}$
$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	1	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778.169 262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.000 293 071 070	41.868	0.0001	$\frac{4.186\ 8}{9.806\ 65}$ = 0.426 934 784	4.186.8
$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 0.001 285 067 46	1	$\frac{453.592\ 37 \times 30.48}{3.6 \times 10^{13}/980.665}$ = 3.766 160 97 × 10 <sup>-7</sup>	$\frac{30.48 \times 980.665 \times 9/5}{10^6}$ = 0.053 803 204 6	$\frac{980.665 \times 30.48}{2\ 326 \times 10^7}$ = 1.285 067 46 × 10 <sup>-6</sup>	$\frac{30.48 \times 10^{-5} \times 9/5}{0.000\ 548\ 64}$ = 0.000 548 64	$\frac{980.665 \times 30.48 \times 10^{-7}}{5/9}$ = 0.005 380 320 46
$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{3\ 600\ 000}{2.326 \times 453.592\ 37}$ = 3 412.141 63	$\frac{3.6 \times 10^{13} / 980.665}{453.592\ 37 \times 30.48}$ = 2 655 223.73	1	$\frac{3.6 \times 10^7 \times 9/5}{453.592\ 37}$ = 142 859.546	$\frac{3\ 600\ 000}{2\ 326 \times 453.592\ 37}$ = 3.412 141 63	$\frac{3.6 \times 10^8 \times 9/5}{980.665 \times 453.592\ 37}$ = 1 456 761 95	$\frac{3.6 \times 10^6 \times 9/5}{453.592\ 37}$ = 14 285.954 6
$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592\ 37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41\ 868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066\ 5}$ = 0.010 197 1621	0.1
$\frac{\text{kcal}}{\text{g} \times \text{K}}$	1000	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778 169.262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.293 071 070	41 868	1	$\frac{4\ 186.8}{9.806\ 65}$ = 426.934 784	4 186.8
$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{9.806\ 65}{4.186\ 8}$ = 2.342 278 11	$\frac{10^5}{30.48 \times 9/5}$ = 1 822.688 83	$\frac{980.665 \times 453.592\ 37}{3.6 \times 10^8 \times 9/5}$ = 0.000 686 453 953	98.066 5	$\frac{9.806\ 65}{4\ 186.8}$ = 0.002 342 278 11	1	9.806 65
$\frac{\text{kJ}}{\text{kg} \times \text{K}}$	$\frac{1.0}{4.186\ 8}$ = 0.238 845 897	$\frac{10^7 \times 5/9}{980.665 \times 30.48}$ = 185.862 535	$\frac{453.592\ 37}{3.6 \times 10^6 \times 9/5}$ = 6.999 882 25 × 10 <sup>-5</sup>	10	$\frac{1.0}{4\ 186.8}$ = 0.000 238 845 897	$\frac{1.0}{9.806\ 65}$ = 0.101 971 621	1

GENERAL NOTE:  
 All values given in the rational fractions are exact.

Example: 1 Btu / (lbm × R) = 778.169 262 ft × lbf / (lbm × R)

**TABLE 5.5**  
**CONVERSION FACTORS FOR SPECIFIC ENTROPY, SPECIFIC HEAT, AND GAS CONSTANT**  
(energy/mass × temperature)

To obtain → Multiply, by ↗ ↓	$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{\text{kcal}}{\text{g} \times \text{K}}$	$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{\text{kJ}}{\text{kg} \times \text{K}}$
$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	1	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778.169 262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.000 293 071 070	41.868	0.0001	$\frac{4.186\ 8}{9.806\ 65}$ = 0.426 934 784	4.186 8
$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 0.001 285 067 46	1	$\frac{453.592\ 37 \times 30.48}{3.6 \times 10^{13}/980.665}$ = 3.766 160 97 × 10 <sup>-7</sup>	$\frac{30.48 \times 980.665 \times 9/5}{10^6}$ = 0.053 803 204 6	$\frac{980.665 \times 30.48}{2\ 326 \times 10^7}$ = 1.285 067 46 × 10 <sup>-6</sup>	$30.48 \times 10^{-5} \times 9/5$ = 0.000 548 64	$\frac{980.665 \times 30.48 \times 10^{-7}}{5/9}$ = 0.005 380 320 46
$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{3\ 600\ 000}{2.326 \times 453.592\ 37}$ = 3 412.141 63	$\frac{3.6 \times 10^{13}/980.665}{453.592\ 37 \times 30.48}$ = 2 655 223.73	1	$\frac{3.6 \times 10^7 \times 9/5}{453.592\ 37}$ = 142 859.546	$\frac{3\ 600\ 000}{2\ 326 \times 453.592\ 37}$ = 3.412 141 63	$\frac{3.6 \times 10^8 \times 9/5}{980.665 \times 453.592\ 37}$ = 1 456 761 95	$\frac{3.6 \times 10^6 \times 9/5}{453.592\ 37}$ = 14 285.954 6
$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592\ 37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41\ 868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066\ 5}$ = 0.010 197 1621	0.1
$\frac{\text{kcal}}{\text{g} \times \text{K}}$	1000	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778 169.262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.293 071 070	41 868	1	$\frac{4\ 186.8}{9.806\ 65}$ = 426.934 784	4 186.8
$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{9.806\ 65}{4.186\ 8}$ = 2.342 278 11	$\frac{10^5}{30.48 \times 9/5}$ = 1 822.688 83	$\frac{980.665 \times 453.592\ 37}{3.6 \times 10^8 \times 9/5}$ = 0.000 686 453 953	98.066 5	$\frac{9.806\ 65}{4\ 186.8}$ = 0.002 342 278 11	1	9.806 65
$\frac{\text{kJ}}{\text{kg} \times \text{K}}$	$\frac{1.0}{4.186\ 8}$ = 0.238 845 897	$\frac{10^7 \times 5/9}{980.665 \times 30.48}$ = 185.862 535	$\frac{453.592\ 37}{3.6 \times 10^6 \times 9/5}$ = 6.999 882 25 × 10 <sup>-5</sup>	10	$\frac{1.0}{4\ 186.8}$ = 0.000 238 845 897	$\frac{1.0}{9.806\ 65}$ = 0.101 971 621	1

GENERAL NOTE:  
All values given in the rational fractions are exact.

Example: 1 Btu / (lbm × R) = 778.169 262 ft × lbf / (lbm × R)

**TABLE 5.5**  
**CONVERSION FACTORS FOR SPECIFIC ENTROPY, SPECIFIC HEAT, AND GAS CONSTANT**  
 (energy/mass × temperature)

To obtain → Multiply, by ↗ ↓	$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{\text{kcal}}{\text{g} \times \text{K}}$	$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{\text{kJ}}{\text{kg} \times \text{K}}$
$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	1	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778.169 262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.000 293 071 070	41.868	0.0001	$\frac{4.186\ 8}{9.806\ 65}$ = 0.426 934 784	4.186 8
$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 0.001 285 067 46	1	$\frac{453.592\ 37 \times 30.48}{3.6 \times 10^{13}/980.665}$ = 3.766 160 97 × 10 <sup>-7</sup>	$\frac{30.48 \times 980.665 \times 9/5}{10^6}$ = 0.053 803 204 6	$\frac{980.665 \times 30.48}{2\ 326 \times 10^7}$ = 1.285 067 46 × 10 <sup>-6</sup>	$\frac{30.48 \times 10^{-5} \times 9/5}{0.000\ 548\ 64}$ = 0.000 548 64	$\frac{980.665 \times 30.48 \times 10^{-7}}{5/9}$ = 0.005 380 320 46
$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{3\ 600\ 000}{2.326 \times 453.592\ 37}$ = 3 412.141 63	$\frac{3.6 \times 10^{13}/980.665}{453.592\ 37 \times 30.48}$ = 2 655 223.73	1	$\frac{3.6 \times 10^7 \times 9/5}{453.592\ 37}$ = 142 859.546	$\frac{3\ 600\ 000}{2\ 326 \times 453.592\ 37}$ = 3.412 141 63	$\frac{3.6 \times 10^8 \times 9/5}{980.665 \times 453.592\ 37}$ = 1 456 761 95	$\frac{3.6 \times 10^6 \times 9/5}{453.592\ 37}$ = 14 285.954 6
$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592\ 37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41\ 868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066\ 5}$ = 0.010 197 1621	0.1
$\frac{\text{kcal}}{\text{g} \times \text{K}}$	1000	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778 169.262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.293 071 070	41 868	1	$\frac{4\ 186.8}{9.806\ 65}$ = 426.934 784	4 186.8
$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{9.806\ 65}{4.186\ 8}$ = 2.342 278 11	$\frac{10^5}{30.48 \times 9/5}$ = 1 822.688 83	$\frac{980.665 \times 453.592\ 37}{3.6 \times 10^8 \times 9/5}$ = 0.000 686 453 953	98.066 5	$\frac{9.806\ 65}{4\ 186.8}$ = 0.002 342 278 11	1	9.806 65
$\frac{\text{kJ}}{\text{kg} \times \text{K}}$	$\frac{1.0}{4.186\ 8}$ = 0.238 845 897	$\frac{10^7 \times 5/9}{980.665 \times 30.48}$ = 185.862 535	$\frac{453.592\ 37}{3.6 \times 10^6 \times 9/5}$ = 6.999 882 25 × 10 <sup>-5</sup>	10	$\frac{1.0}{4\ 186.8}$ = 0.000 238 845 897	$\frac{1.0}{9.806\ 65}$ = 0.101 971 621	1

GENERAL NOTE:  
 All values given in the rational fractions are exact.

Example: 1 Btu / (lbm × R) = 778.169 262 ft × lbf / (lbm × R)

**TABLE 5.5**  
**CONVERSION FACTORS FOR SPECIFIC ENTROPY, SPECIFIC HEAT, AND GAS CONSTANT**  
(energy/mass × temperature)

To obtain → Multiply, by ↗ ↓	$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{\text{kcal}}{\text{g} \times \text{K}}$	$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{\text{kJ}}{\text{kg} \times \text{K}}$
$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	1	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778.169 262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.000 293 071 070	41.868	0.0001	$\frac{4.186\ 8}{9.806\ 65}$ = 0.426 934 784	4.186.8
$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 0.001 285 067 46	1	$\frac{453.592\ 37 \times 30.48}{3.6 \times 10^{13}/980.665}$ = 3.766 160 97 × 10 <sup>-7</sup>	$\frac{30.48 \times 980.665 \times 9/5}{10^6}$ = 0.053 803 204 6	$\frac{980.665 \times 30.48}{2\ 326 \times 10^7}$ = 1.285 067 46 × 10 <sup>-6</sup>	$\frac{30.48 \times 10^{-5} \times 9/5}{0.000\ 548\ 64}$ = 0.000 548 64	$\frac{980.665 \times 30.48 \times 10^{-7}}{5/9}$ = 0.005 380 320 46
$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{3\ 600\ 000}{2.326 \times 453.592\ 37}$ = 3 412.141 63	$\frac{3.6 \times 10^{13} / 980.665}{453.592\ 37 \times 30.48}$ = 2 655 223.73	1	$\frac{3.6 \times 10^7 \times 9/5}{453.592\ 37}$ = 142 859.546	$\frac{3\ 600\ 000}{2\ 326 \times 453.592\ 37}$ = 3.412 141 63	$\frac{3.6 \times 10^8 \times 9/5}{980.665 \times 453.592\ 37}$ = 1 456 761 95	$\frac{3.6 \times 10^6 \times 9/5}{453.592\ 37}$ = 14 285.954 6
$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592\ 37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41\ 868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066\ 5}$ = 0.010 197 1621	0.1
$\frac{\text{kcal}}{\text{g} \times \text{K}}$	1000	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778 169.262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.293 071 070	41 868	1	$\frac{4\ 186.8}{9.806\ 65}$ = 426.934 784	4 186.8
$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{9.806\ 65}{4.186\ 8}$ = 2.342 278 11	$\frac{10^5}{30.48 \times 9/5}$ = 1 822.688 83	$\frac{980.665 \times 453.592\ 37}{3.6 \times 10^8 \times 9/5}$ = 0.000 686 453 953	98.066 5	$\frac{9.806\ 65}{4\ 186.8}$ = 0.002 342 278 11	1	9.806 65
$\frac{\text{kJ}}{\text{kg} \times \text{K}}$	$\frac{1.0}{4.186\ 8}$ = 0.238 845 897	$\frac{10^7 \times 5/9}{980.665 \times 30.48}$ = 185.862 535	$\frac{453.592\ 37}{3.6 \times 10^6 \times 9/5}$ = 6.999 882 25 × 10 <sup>-5</sup>	10	$\frac{1.0}{4\ 186.8}$ = 0.000 238 845 897	$\frac{1.0}{9.806\ 65}$ = 0.101 971 621	1

GENERAL NOTE:  
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Example: 1 Btu / (lbm × R) = 778.169 262 ft × lbf / (lbm × R)

**TABLE 5.5**  
**CONVERSION FACTORS FOR SPECIFIC ENTROPY, SPECIFIC HEAT, AND GAS CONSTANT**  
 (energy/mass × temperature)

To obtain → Multiply, by ↗ ↓	$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{\text{kcal}}{\text{g} \times \text{K}}$	$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{\text{kJ}}{\text{kg} \times \text{K}}$
$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	1	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778.169 262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.000 293 071 070	41.868	0.0001	$\frac{4.186\ 8}{9.806\ 65}$ = 0.426 934 784	4.186 8
$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 0.001 285 067 46	1	$\frac{453.592\ 37 \times 30.48}{3.6 \times 10^{13}/980.665}$ = 3.766 160 97 × 10 <sup>-7</sup>	$\frac{30.48 \times 980.665 \times 9/5}{10^6}$ = 0.053 803 204 6	$\frac{980.665 \times 30.48}{2\ 326 \times 10^7}$ = 1.285 067 46 × 10 <sup>-6</sup>	$\frac{30.48 \times 10^{-5} \times 9/5}{0.000\ 548\ 64}$ = 0.000 548 64	$\frac{980.665 \times 30.48 \times 10^{-7}}{5/9}$ = 0.005 380 320 46
$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{3\ 600\ 000}{2.326 \times 453.592\ 37}$ = 3 412.141 63	$\frac{3.6 \times 10^{13}/980.665}{453.592\ 37 \times 30.48}$ = 2 655 223.73	1	$\frac{3.6 \times 10^7 \times 9/5}{453.592\ 37}$ = 142 859.546	$\frac{3\ 600\ 000}{2\ 326 \times 453.592\ 37}$ = 3.412 141 63	$\frac{3.6 \times 10^8 \times 9/5}{980.665 \times 453.592\ 37}$ = 1 456 761 95	$\frac{3.6 \times 10^6 \times 9/5}{453.592\ 37}$ = 14 285.954 6
$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592\ 37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41\ 868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066\ 5}$ = 0.010 197 1621	0.1
$\frac{\text{kcal}}{\text{g} \times \text{K}}$	1000	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778 169.262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.293 071 070	41 868	1	$\frac{4\ 186.8}{9.806\ 65}$ = 426.934 784	4 186.8
$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{9.806\ 65}{4.186\ 8}$ = 2.342 278 11	$\frac{10^5}{30.48 \times 9/5}$ = 1 822.688 83	$\frac{980.665 \times 453.592\ 37}{3.6 \times 10^8 \times 9/5}$ = 0.000 686 453 953	98.066 5	$\frac{9.806\ 65}{4\ 186.8}$ = 0.002 342 278 11	1	9.806 65
$\frac{\text{kJ}}{\text{kg} \times \text{K}}$	$\frac{1.0}{4.186\ 8}$ = 0.238 845 897	$\frac{10^7 \times 5/9}{980.665 \times 30.48}$ = 185.862 535	$\frac{453.592\ 37}{3.6 \times 10^6 \times 9/5}$ = 6.999 882 25 × 10 <sup>-5</sup>	10	$\frac{1.0}{4\ 186.8}$ = 0.000 238 845 897	$\frac{1.0}{9.806\ 65}$ = 0.101 971 621	1

GENERAL NOTE:  
 All values given in the rational fractions are exact.

Example: 1 Btu / (lbm × R) = 778.169 262 ft × lbf / (lbm × R)

**TABLE 5.5**  
**CONVERSION FACTORS FOR SPECIFIC ENTROPY, SPECIFIC HEAT, AND GAS CONSTANT**  
(energy/mass × temperature)

To obtain → Multiply, by ↗ ↓	$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{\text{kcal}}{\text{g} \times \text{K}}$	$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{\text{kJ}}{\text{kg} \times \text{K}}$
$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	1	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778.169 262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.000 293 071 070	41.868	0.0001	$\frac{4.186\ 8}{9.806\ 65}$ = 0.426 934 784	4.186 8
$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 0.001 285 067 46	1	$\frac{453.592\ 37 \times 30.48}{3.6 \times 10^{13}/980.665}$ = 3.766 160 97 × 10 <sup>-7</sup>	$\frac{30.48 \times 980.665 \times 9/5}{10^6}$ = 0.053 803 204 6	$\frac{980.665 \times 30.48}{2\ 326 \times 10^7}$ = 1.285 067 46 × 10 <sup>-6</sup>	$30.48 \times 10^{-5} \times 9/5$ = 0.000 548 64	$\frac{980.665 \times 30.48 \times 10^{-7}}{5/9}$ = 0.005 380 320 46
$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{3\ 600\ 000}{2.326 \times 453.592\ 37}$ = 3 412.141 63	$\frac{3.6 \times 10^{13} / 980.665}{453.592\ 37 \times 30.48}$ = 2 655 223.73	1	$\frac{3.6 \times 10^7 \times 9/5}{453.592\ 37}$ = 142 859.546	$\frac{3\ 600\ 000}{2\ 326 \times 453.592\ 37}$ = 3.412 141 63	$\frac{3.6 \times 10^8 \times 9/5}{980.665 \times 453.592\ 37}$ = 1 456 761 95	$\frac{3.6 \times 10^6 \times 9/5}{453.592\ 37}$ = 14 285.954 6
$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592\ 37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41\ 868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066\ 5}$ = 0.010 197 1621	0.1
$\frac{\text{kcal}}{\text{g} \times \text{K}}$	1000	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778 169.262	$\frac{2.326 \times 453.592\ 37}{3\ 600\ 000}$ = 0.293 071 070	41 868	1	$\frac{4\ 186.8}{9.806\ 65}$ = 426.934 784	4 186.8
$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{9.806\ 65}{4.186\ 8}$ = 2.342 278 11	$\frac{10^5}{30.48 \times 9/5}$ = 1 822.688 83	$\frac{980.665 \times 453.592\ 37}{3.6 \times 10^8 \times 9/5}$ = 0.000 686 453 953	98.066 5	$\frac{9.806\ 65}{4\ 186.8}$ = 0.002 342 278 11	1	9.806 65
$\frac{\text{kJ}}{\text{kg} \times \text{K}}$	$\frac{1.0}{4.186\ 8}$ = 0.238 845 897	$\frac{10^7 \times 5/9}{980.665 \times 30.48}$ = 185.862 535	$\frac{453.592\ 37}{3.6 \times 10^6 \times 9/5}$ = 6.999 882 25 × 10 <sup>-5</sup>	10	$\frac{1.0}{4\ 186.8}$ = 0.000 238 845 897	$\frac{1.0}{9.806\ 65}$ = 0.101 971 621	1

GENERAL NOTE:  
All values given in the rational fractions are exact.

Example: 1 Btu / (lbm × R) = 778.169 262 ft × lbf / (lbm × R)

**TABLE 5.5**  
**CONVERSION FACTORS FOR SPECIFIC ENTROPY, SPECIFIC HEAT, AND GAS CONSTANT**  
(energy/mass × temperature)

To obtain → Multiply, by ↗ ↓	$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{\text{kcal}}{\text{g} \times \text{K}}$	$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{\text{kJ}}{\text{kg} \times \text{K}}$
$\frac{\text{Btu}}{\text{lbm} \times \text{R}}$	1	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778.169 262	$\frac{2.326 \times 453.592.37}{3.600.000}$ = 0.000 293 071 070	41.868	0.0001	$\frac{4.186.8}{9.806.65}$ = 0.426 934 784	4.186.8
$\frac{\text{ft} \times \text{lb}_f}{\text{lbm} \times \text{R}}$	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 0.001 285 067 46	1	$\frac{453.592.37 \times 30.48}{3.6 \times 10^{13} / 980.665}$ = 3.766 160 97 × 10 <sup>-7</sup>	$\frac{30.48 \times 980.665 \times 9/5}{10^6}$ = 0.053 803 204 6	$\frac{980.665 \times 30.48}{2.326 \times 10^7}$ = 1.285 067 46 × 10 <sup>-6</sup>	$30.48 \times 10^{-5} \times 9/5$ = 0.000 548 64	$\frac{980.665 \times 30.48 \times 10^{-7}}{5/9}$ = 0.005 380 320 46
$\frac{\text{kw} \times \text{hr}}{\text{lbm} \times \text{R}}$	$\frac{3.600.000}{2.326 \times 453.592.37}$ = 3 412.141 63	$\frac{3.6 \times 10^{13} / 980.665}{453.592.37 \times 30.48}$ = 2 655 223.73	1	$\frac{3.6 \times 10^7 \times 9/5}{453.592.37}$ = 142 859.546	$\frac{3.600.000}{2.326 \times 453.592.37}$ = 3.412 141 63	$\frac{3.6 \times 10^8 \times 9/5}{980.665 \times 453.592.37}$ = 1 456 761 95	$\frac{3.6 \times 10^6 \times 9/5}{453.592.37}$ = 14 285.954 6
$\frac{\text{bar} \times \text{cm}^3}{\text{g} \times \text{K}}$	$\frac{1.0}{41.868}$ = 0.023 884 589 7	$\frac{10^6}{30.48 \times 980.665 \times 9/5}$ = 18.586 253 5	$\frac{453.592.37}{3.6 \times 10^7 \times 9/5}$ = 6.999 882 25 × 10 <sup>-6</sup>	1	$\frac{1.0}{41.868}$ = 2.388 458 97 × 10 <sup>-5</sup>	$\frac{1.0}{98.066.5}$ = 0.010 197 1621	0.1
$\frac{\text{kcal}}{\text{g} \times \text{K}}$	1000	$\frac{2.326 \times 10^7}{980.665 \times 30.48}$ = 778 169.262	$\frac{2.326 \times 453.592.37}{3.600.000}$ = 0.293 071 070	41 868	1	$\frac{4.186.8}{9.806.65}$ = 426.934 784	4 186.8
$\frac{\text{kp} \times \text{m}}{\text{g} \times \text{K}}$	$\frac{9.806.65}{4.186.8}$ = 2.342 278 11	$\frac{10^5}{30.48 \times 9/5}$ = 1 822.688 83	$\frac{980.665 \times 453.592.37}{3.6 \times 10^8 \times 9/5}$ = 0.000 686 453 953	98.066 5	$\frac{9.806.65}{4.186.8}$ = 0.002 342 278 11	1	9.806 65
$\frac{\text{kJ}}{\text{kg} \times \text{K}}$	$\frac{1.0}{4.186.8}$ = 0.238 845 897	$\frac{10^7 \times 5/9}{980.665 \times 30.48}$ = 185.862 535	$\frac{453.592.37}{3.6 \times 10^6 \times 9/5}$ = 6.999 882 25 × 10 <sup>-5</sup>	10	$\frac{1.0}{4.186.8}$ = 0.000 238 845 897	$\frac{1.0}{9.806.65}$ = 0.101 971 621	1

GENERAL NOTE:  
All values given in the rational fractions are exact.

Example: 1 Btu / (lbm × R) = 778.169 262 ft × lbf / (lbm × R)